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Synthesis of 2',3'-Dideoxy-6',6'-difluorocarbocyclic Nucleosides

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ABSTRACT

2',3'-Dideoxy-6',6'-difluorouracils, a novel series of *gem*-difluoromethylenated carbocyclic nucleosides, were synthesized from (*Z*)-but-2-ene-1,4-diol in 14 steps. A notable step was the construction of the carbocyclic ring via ring-closing metathesis and the incorporation of *gem*-difluoromethylene group by way of silicon-induced Reformatskii—Claisen reaction of chlorodifluoroacetic ester 3.

In recent years, attention has been increasingly focused on structural modifications of carbocyclic nucleosides. Due to the absence of a glycosidic linkage, carbocyclic nucleosides are chemically more stable and not subject to the phosphorylases that cleave the N-glycosidic linkage in conventional nucleosides. Many carbocyclic nucleosides have now been identified to exhibit antiviral and antitumor activities. Abacavir (Figure 1) has been used as an anti-HIV agent. Entecavir, a carbocyclic nucleoside with an exocyclic double bond, is undergoing phase III clinical trials for the treatment of chronic hepatitis B virus infection. 6'-Fluorocarbocyclic

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nucleoside **1** exhibited moderate activities against herpes simplex virus type 1 (HSV-1) and type 2 (HSV-2) in vitro.⁵ The 2',3'-dideoxynucleosides (ddNs) have been proved to be the most effective therapeutic agents against human

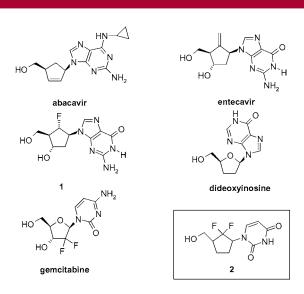


Figure 1. Rationale for the design of the target nucleosides 2.

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immunodeficiency virus (HIV) and hepatitis B virus (HBV). Among them, dideoxyinosine (DDI) had been developed into an anti-HIV drug. The *gem*-difluoromethylene (CF₂) group has been suggested by Blackburn as an isopolar and isosteric substituent for oxygen. Since then, the CF₂ group was used extensively to modify nucleoside analogues. For example, 2'-deoxy-2',2'-difluorocytidine (gemcitabine) has been approved as a drug for solid tumor treatment. On the basis of the above consideration and our ongoing efforts to develop new antiviral and anticancer agents, we designed the 2',3'-dideoxycarbocyclic nucleosides 2 (Figure 1), a new type of analogue of DDI, by replacing the oxygen with difluoromethylene group (CF₂) based on the bioisosteric rationale. Herein, an efficient route to synthesize 2',3'-dideoxycarbocyclic nucleosides 2 is described.

As illustrated in Scheme 1, retrosynthetic analysis showed that the target nucleosides could be synthesized from cyclic

amine 10, which could be used to introduce a base moiety at the C1 position using a well-known procedure. However, construction of the special backbone of 10, especially introduction of *gem*-difluoromethylene to the C4 position of 10, is very difficult. Although DAST appears to be the most common reagent for introduction of a *gem*-difluoromethylene group, very few sterically hindered five-membered cyclic ketones have been difluorinated by DAST. We envisioned that compound 6 could be converted into 10 via ring-closing metathesis. Compound 6 can be derived from chlorodifluoroacetic ester 3 through Reformatskii—Claisen reaction.

The synthesis of the nucleosides **2** began with (*Z*)-2-butene-1,4-diol (Scheme 2). Protection of one of its hydroxy

groups, follwed by esterification of another one with chlorodifluoroacetic acid catalyzed by sulfonic acid, gave 3 in multigram quantities.^{9,10} Then, **3** underwent a silicon-induced Reformatskii-Claisen reaction¹¹ when a mixture of 3, chlorotrimethylsilane, and freshly activated zinc dust was heated for 20 h in dry acetonitrile at 100 °C. The resulting crude product was esterified with ethanol to give 4 in 84% yield (two steps). Ester 4 was then transformed into Weinreb amide 5 in 85% yield. Treatment of 5 with allylmagnesium bromide resulted in successful conversion into the corresponding β, γ -unsaturated ketone, which was transformed to 6 in 91% yield (two steps) by double-bond isomerization. With compound 6 in hand, we turned our attention to the ring-closing metathesis (RCM) of compound 6. Initially, the RCM of 6 was carried out in the presence of the firstgeneration Grubbs' catalyst, and the reaction did not occur. Fortunately, when compound 6 was subjected to the secondgeneration Grubbs' catalyst¹² in refluxing toluene, the reaction resulted in complete conversion and compound 7 was isolated in 98% yield. Ketone 7 was transformed to alcohols 8a and 8b via Luche reduction¹³ in a 2.9:1 cis/trans ratio, which can be separated easily through column chromatography. The relative configuration of 8a was confirmed by the structure of 2a, which was identified by X-ray analysis.

Hydrogenation of **8a** with the catalyst of Pd/C in benzene for 24 h gave compound **9a** in 84% yield (Scheme 3).¹⁴ Treatment of alcohol **9a** with trifluoromethanesulfonic

4258 Org. Lett., Vol. 6, No. 23, 2004

3521

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anhydride and pyridine followed by substitution reaction with sodium azide in DMF gave the azide compound, which was directly reduced by hydrogenation to give cyclic amine **10a** in 56% yield (three steps). The construction of pyrimidine was followed by the reported procedure. Condensation of cyclic amine **10a** with 3-ethoxy-2-propenoyl isocyanate in DMF at -25 °C followed by ring closure with 2 N sulfuric acid in dioxane successfully produced the nucleoside. Removal of the benzyl group by hydrogenation gave the target molecule **2a** in 58% yield (three steps). The relative configuration of racemic **2a** was determined by X-ray crystal structure (Figure 2). With the same synthetic route, isomer **2b** was prepared from **8b**.

In conclusion, we have completed a 14-step synthesis of racemic 2',3'-dideoxy-6',6'-difluorocarbocyclic nucleoside 2a in 7.6% overall yield and 2b in 1.5% overall yield.

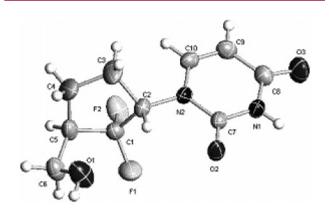


Figure 2. ORTEP drawing of the X-ray crystallographic structure of **2a**.

Reformatskii—Claisen rearrangement and ring-closing metathesis are the key steps of the synthesis. The flexibility in our approach to access these novel nucleosides is noteworthy, and we are currently investigating enantioselective synthesis of 2',3'-dideoxy-6',6'-difluoro-carbocyclic nucleosides.

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Supporting Information Available: Experimental procedures and characterization data for all new compounds and crystallographic data for compound **2a** (CIF). This material is available free of charge via the Internet at http://pubs.acs.org.

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Org. Lett., Vol. 6, No. 23, **2004**

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